

the atomic numbering scheme used. The figure was designed by *KPLOT* (Hundt, 1979) and executed with *ORTEP* (Johnson, 1965).

Lists of structure factors, anisotropic displacement parameters, H-atom coordinates and complete geometry have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 55771 (48 pp.). Copies may be obtained through The Technical Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England. [CIF reference: SH1008]

References

- Domingos, A. M. & Sheldrick, G. M. (1974). *Acta Cryst.* B32, 519–521.
 Flack, H. D. (1983). *Acta Cryst.* A39, 876–881.
 Hundt, R. (1979). *KPLOT*. Program for drawing and studying crystal structures. Univ. of Bonn, Germany.
 Johnson, C. K. (1965). *ORTEP*. Report ORNL-3794. Oak Ridge National Laboratory, Tennessee, USA. [Modified for local use by R. Hundt (1969).]
 Puff, H. & Reuter, H. (1989). *J. Organomet. Chem.* 379, 223–234.
 Sheldrick, G. M. (1986). *SHELXS86*. Program for crystal structure determination. Univ. of Göttingen, Germany.
 Sheldrick, G. M. (1992). *SHELXL92*. Program for crystal structure determination. Univ. of Göttingen, Germany.

Acta Cryst. (1993). C49, 956–957

Bis(5-bromocytosinium) Tetrachloropalladate(II)

U. CASELLATO

Istituto di Chimica e Tecnologia dei Radioelementi, CNR, Padua, Italy

R. ETTORRE AND R. GRAZIANI

Dipartimento di Chimica Inorganica, Metallorganica e Analitica, Università di Padova, Padua, Italy

(Received 15 June 1992; accepted 26 October 1992)

Abstract

The structure consists of discrete $C_4H_5BrN_3O^+$ and $PdCl_4^{2-}$ ions in a network of N—H...Cl hydrogen bonds with the crystal packing being different from that of the analogous compound bis(cytosinium) tetrachloropalladate(II). The largest deviations from bond lengths and angles of 5-bromocytosine [4-amino-5-bromo-2(1*H*)-pyrimidinone] observed upon protonation are analogous to those found for cytosine.

Comment

This report extends work on halogenated nitrogen bases (Valle & Ettore, 1992; Casellato, Graziani & Sánchez González, 1992).

The structure consists of discrete $C_4H_5BrN_3O^+$ and $PdCl_4^{2-}$ ions in a network of N—H...Cl hydrogen bonds. The $PdCl_4^{2-}$ anion is planar with the metal atom located at the origin of the centrosymmetric triclinic cell. Coordination bond lengths and angles are normal. The largest deviations from bond lengths and angles of 5-bromocytosine (Kato, Takenaka & Sasada, 1979) observed upon protonation are the decrease of C(4)—N(4) (*ca* 0.04 Å, 4σ), C(2)—O(2) (*ca* 0.05 Å, 4σ), N(3)—C(4)—C(5) (*ca* 4°, 5σ) and N(1)—C(2)—N(3) (*ca* 5°, 6σ), and the increase of C(2)—N(3)—C(4) (*ca* 6°, 8σ) [$\sigma = (\sigma_1^2 + \sigma_2^2)^{1/2}$]. Analogous effects have been observed for cytosine (Bagieu-Beucher, 1990).

The crystal packing is different from that of bis(cytosinium) tetrachloropalladate(II) (Kindberg & Amma, 1975) where, for example, the planes of the anions have two orientations.

The O(2)...Br (*x*, *y* - 1, *z*) contact of 2.972 (4) Å is remarkably close in relation to reported van der Waals radii for O (1.54 Å) and Br atoms (1.54–1.84 Å) (Nyburg & Faerman, 1985).

As shown in Table 2, each coordinated Cl(1) atom forms hydrogen bonds with the H—N(1) groups of two different cations, and Cl(2) is involved in bifurcated hydrogen bonds with the H—N(3) and H—N(4)H groups of a third cation, so that each centrosymmetric $PdCl_4^{2-}$ anion is hydrogen bonded to six cations, which gives compactness and stability to the crystal packing.

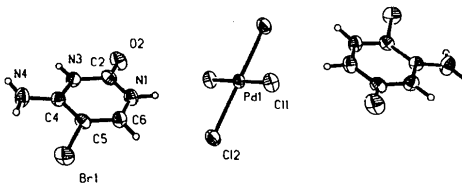


Fig. 1. Plot of the anion and two cations of the title compound showing the atomic numbering scheme.

Experimental

Crystal data

$[C_4H_5BrN_3O]_2[PdCl_4]$

$M_r = 630.2$

Triclinic

$P\bar{1}$

$a = 9.146$ (4) Å

$b = 8.650$ (4) Å

$c = 6.622$ (3) Å

$\alpha = 109.05$ (4)°

$\beta = 71.05$ (4)°

$\gamma = 109.03$ (5)°

Mo $K\alpha$ radiation

$\lambda = 0.7107$ Å

Cell parameters from 30 reflections

$\theta = 6$ –12°

$\mu = 5.987$ mm⁻¹

Room temperature

Plates

0.2 × 0.2 × 0.2 mm

Brick red

$V = 453.9$ (4) Å³
 $Z = 1$
 $D_x = 2.305$ Mg m⁻³

Crystal source: chemical
 synthesis

the electron density map and refined with a fixed temperature factor ($U_{iso} = 0.07$ Å²). All calculations were performed on a MicroVAX using *SHELX76* (Sheldrick, 1976)

Data collection

Philips PW1100 diffractometer
 $\theta/2\theta$ scans
 Absorption correction: empirical
 $T_{min} = 0.890$, $T_{max} = 0.995$
 4279 measured reflections
 4001 independent reflections

2435 observed reflections [$F > 12\sigma(F)$]
 $\theta_{max} = 25^\circ$
 $h = -13 \rightarrow 14$
 $k = -13 \rightarrow 13$
 $l = 0 \rightarrow 10$
 2 standard reflections monitored every 100 reflections
 intensity variation: none

Refinement

Refinement on F
 Final $R = 0.031$
 $wR = 0.031$
 $S = 0.777$

$w = 1$
 $(\Delta/\sigma)_{max} = 0.02$
 $\Delta\rho_{max} = 1.2$ e Å⁻³

2435 reflections
 121 parameters
 Only coordinates of H atoms refined

Atomic scattering factors from *International Tables for X-ray Crystallography* (1974, Vol. IV)

Table 1. Fractional atomic coordinates and equivalent isotropic thermal parameters (Å²)

U_{eq} is defined as one third of the trace of the orthogonalized U_{ij} tensor.

	x	y	z	U_{eq}
Pd	0.0	0.0	0.0	0.0255 (1)
Cl(1)	-0.1585 (1)	0.1833 (1)	0.2099 (2)	0.0357 (4)
Cl(2)	0.1108 (1)	0.1579 (1)	-0.2580 (2)	0.0417 (4)
N(1)	0.4893 (4)	0.1183 (4)	-0.2567 (6)	0.034 (1)
C(2)	0.6059 (4)	0.0570 (4)	-0.2453 (6)	0.032 (1)
N(3)	0.7499 (4)	0.1744 (4)	-0.2416 (6)	0.032 (1)
C(4)	0.7826 (4)	0.3370 (4)	-0.2489 (6)	0.029 (1)
C(5)	0.6573 (4)	0.3896 (4)	-0.2686 (6)	0.029 (1)
C(6)	0.5147 (4)	0.2776 (5)	-0.2725 (7)	0.033 (1)
O(2)	0.5860 (4)	-0.0862 (3)	-0.2390 (6)	0.047 (1)
N(4)	0.9248 (4)	0.4340 (5)	-0.2375 (7)	0.044 (2)
Br	0.69082 (6)	0.60824 (5)	-0.29450 (8)	0.0449 (2)

Table 2. Geometric parameters (Å, °)

Pd—Cl(1)	2.302 (2)	C(4)—C(5)	1.419 (7)
N(1)—C(2)	1.368 (7)	C(6)—N(1)	1.354 (6)
N(3)—C(4)	1.353 (5)	C(4)—N(4)	1.309 (5)
C(5)—C(6)	1.349 (5)		
C(2)—O(2)	1.205 (5)	N(1)···Cl(1) ⁱ	3.291 (4)
C(5)—Br	1.871 (4)	N(3)···Cl(2) ⁱⁱ	3.316 (4)
Pd—Cl(2)	2.304 (2)	N(4)···Cl(2) ⁱⁱⁱ	3.302 (5)
C(2)—N(3)	1.378 (5)	N(4)···Cl(1) ⁱⁱⁱ	3.287 (4)
Cl(1)—Pd—Cl(2)	90.3 (1)	N(1)—C(2)—N(3)	113.9 (3)
C(2)—N(3)—C(4)	126.3 (4)	N(3)—C(4)—C(5)	116.5 (4)
C(4)—C(5)—C(6)	118.6 (4)	C(5)—C(6)—N(1)	121.5 (4)
C(6)—N(1)—C(2)	123.1 (4)	N(1)—C(2)—O(2)	123.6 (4)
O(2)—C(2)—N(3)	122.5 (4)	N(3)—C(4)—N(4)	118.8 (4)
N(4)—C(4)—C(5)	124.7 (4)	C(4)—C(5)—Br	120.6 (3)
Br—C(5)—C(6)	120.9 (3)		

Symmetry codes: (i) $-x, -y, -z$; (ii) $1 + x, y, z$; (iii) $1 - x, 1 - y, -z$.

Intensities were corrected for Lp and for absorption (North Phillips & Mathews, 1968). The structure was solved by the heavy-atom method and final scale factors, atomic coordinates and anisotropic thermal parameters were obtained by full-matrix least-squares refinement. H-atom positions were obtained from

Lists of structure factors, anisotropic thermal parameters, H-atom coordinates and possible hydrogen bonds have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 55789 (18 pp.). Copies may be obtained through The Technical Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England. [CIF reference: NA1012]

References

- Bagieu-Beucher, M. (1990). *Acta Cryst.* **C46**, 238–240.
 Casellato, U., Graziani, R. & Sánchez González, A. (1992). *Acta Cryst.* **C48**, 2125–2126.
 Kato, M., Takenaka, A. & Sasada, Y. (1979). *Bull. Chem. Soc. Jpn.* **52**, 49–51.
 Kindberg, B. L. & Amma, E. L. (1975). *Acta Cryst.* **B31**, 1492–1494.
 North, A. C. T., Phillips, D. C. & Mathews, F. S. (1968). *Acta Cryst.* **A24**, 351–359.
 Nyburg, S. C. & Faerman, C. H. (1985). *Acta Cryst.* **B41**, 274–279.
 Sheldrick, G. M. (1976). *SHELX76*. Program for crystal structure determination. Univ. of Cambridge, England.
 Valle, G. & Ettore, R. (1992). *Acta Cryst.* **C48**, 919–921.

Acta Cryst. (1993). **C49**, 957–961

2-(η^2 -Dithioformato)-2-(triphenylphosphine)-1,7-dicarbonyl-2-rhoda-closo-dodecaborane(12)

GEORGE FERGUSON

Department of Chemistry and Biochemistry, University of Guelph, Guelph, Ontario, Canada N1G 2W1

SIOBHAN COUGLAN AND TREVOR R. SPALDING

Department of Chemistry, University College Cork, Cork, Ireland

(Received 10 August 1992; accepted 19 November 1992)

Abstract

The asymmetric unit has two independent molecules which have very similar conformations. The dithioformato ligand is bonded asymmetrically with Rh—S distances of 2.355(1) and 2.432(1) Å in one molecule and 2.347(1) and 2.423(1) Å in the other; in each case, the shorter Rh—S bond is *trans* to a C atom and the longer one *trans* to a B atom. The Rh—P bond [2.314(1) and 2.320(1) Å, respectively, in the two molecules] is *trans* to the second C atom in the C₂B₉ cage.